

COMPONENTS OF *Haplophyllum bucharicum*

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Continuing a study of alkaloids of the genus *Haplophyllum* [1], we have investigated the epigeal part of *H. bucharicum* Litv. collected by S. A. Khamidkhodzhaev in the environs of the village of Derbent in the Baisun region of Sukhandar' province in the budding-incipient flowering stage on May 17, 1980. The plants from this growth site have not been studied previously [2].

A methanolic extract obtained from the dry comminuted epigeal part (4300 g) was separated into basic, acidic, and neutral fractions. By chromatography of the neutral fraction on a column of alumina, by rechromatography of the individual fractions on a column of silica gel and crystallization, five substances were isolated: (I) — mp 76°C (benzene-ethyl acetate); (II) — mp 134–135°C (acetone); (III) — mp 176–177°C (methanol); (IV) — mp 113–114°C (acetone); and (V) — mp 150–151°C (methanol).

On the basis of features of its IR, PMR, and mass spectra, substance (I) was identified as n-triacantonol [3]. A comparison of the mass spectrum of (II) with that published by Razakov et al. [4] showed that compound (II) was  $\beta$ -sitosterol ( $M^+$  414) containing as impurities very small amounts of sigmasterol ( $M^+$  412), campesterol ( $M^+$  400), and cholesterol ( $M^+$  386).

Substances (III) and (V) were identified as the alkaloids skimmianine and bucharaine, respectively, which have been isolated previously from this plant obtained from a different growth site [5].

Substance (IV) contains a nitrogen atom. Its UV spectrum in ethanol ( $\lambda_{\text{max}}$  277, 272, 282, 313, 324, 338 nm;  $\log \epsilon$  4.10, 3.50, 3.41, 3.35, 3.46, 3.35) is characteristic for 4-alkoxy-2-quinolones [6] and, like the latter, did not change on acidification. The IR spectrum of (IV) contains absorption bands at 3300 and 1650  $\text{cm}^{-1}$  (NH-CO). Mass spectrum, m/z (%) 297 ( $M^+$ , 6), 229 (M – 68, 25), 228 (M – 69, 100), 214 (18), 212 (27), 200 (16), 186 (26), 174 (24), 69 (36). The NMR spectrum of (IV) ( $\text{CDCl}_3$ ) shows signals at (ppm) 7.65 (quadruplet, 1 H), 7.45–7.08 (multiplet, 3 H) — the aromatic protons  $H_5$  and  $H_{6,7,8}$ , respectively; 5.51 and 5.25 (triplets, 1 H each,  $J$  = 7.5 Hz); 4.44 and 3.37 (doublet, 2 H each,  $J$  = 7.5 Hz); and 1.82, 1.75, 1.61 (singlets, 3 H, 3 H, and 6 H, respectively) — two isopentyl groups, one of which is attached to the 2-quinolone nucleus directly and the other through an oxygen atom. These facts give grounds for suggesting that (IV) is 3-dimethylallyl-4-dimethylallyloxy-2-quinoline, which has been isolated previously from *Haplophyllum tuberculatum* [7]. A chemical confirmation of this was the conversion of (IV) on boiling with 85% formic acid into haplofoline and dihydroflindersine [7]. This is the first time that the alkaloid (IV) has been detected in Central Asian species of plants of the genus *Haplophyllum*.

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BUCHAPINE — A NEW ALKALOID FROM *Haplophyllum bucharicum*

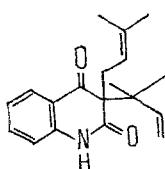
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Continuing a study of the chemical composition of the epigeal part of *Haplophyllum bucharicum* Litv. [1], from the neutral fraction of the methanolic extract, by chromatography on columns of alumina and silica gel, we have obtained a new alkaloid (I) with mp 134–135°C (hexane) and have called it buchapine. Buchapine has the composition C<sub>19</sub>H<sub>23</sub>NO<sub>2</sub>, identical with that of 3-dimethylallyl-4-dimethylallyloxy-2-quinoline (II) [2].

The mass spectrum of (I) — m/z (%) 297 (M<sup>+</sup> 10), 229 (M – 68, 34), 228 (M – 69, 100), 214 (20), 212 (24), 200 (16), 186 (38), 174 (24), 69 (20) — almost coincides with that of (II): There are only slight differences in the intensities of some peaks. These results, and also the presence in the IR spectrum of buchapine of absorption bands at 1692 cm<sup>–1</sup> together with the band of an amide carbonyl group at 1660 cm<sup>–1</sup> have permitted the suggestion that it is based on a 1,2,3,4-tetrahydroquinoline-2,4-dione nucleus with two prenyl substituents in position 3. This conclusion is in harmony with the UV spectrum of (I).

UV spectrum of buchapine:  $\lambda_{\text{max}}$  (nm) 234, 238 (shoulder), 242 infl., 244 infl., 258 shoulder, 324, 329 shoulder, 325 infl. (log ε 4.44, 4.41, 4.32, 3.72, 3.60, 3.41, 3.50, 3.40) differs from those of the 4-alkoxy-2-quinolone alkaloids, in particular (II) [2], by the absence of maxima in the 260–290 nm region characteristic for this group of substances [3], but it is close to those of 3,3-diprenyl-1,2,3,4-tetrahydroquinoline-2,4-dione compounds [4]. The structure of buchapine follows from its PMR spectrum (CDCl<sub>3</sub>, δ scale), which contains signals at 7.74 and 7.5–6.82 ppm (doublet, 1 H, J = 8.5 Hz, and multiplet, 3 H) relating to the H-5 and H-6,7,8 aromatic protons, and also at 5.76 and 4.81 ppm (A<sub>2</sub>B system, 1 H and 2 H respectively: —CH=CH<sub>2</sub>), 4.65 and 2.77 ppm (triplet, 1 H; doublet, 2 H, J = 7.5 Hz, =CH—CH<sub>2</sub>), and 1.88, 1.40, and 1.09 ppm (singlets, 3 H, 3 H, and 6 H, respectively), belonging to the protons of two prenyl substituents one of which has the structure —C(CH<sub>3</sub>)<sub>2</sub>—CH=CH<sub>2</sub> and the other —CH<sub>2</sub>—CH=C(CH<sub>3</sub>)<sub>2</sub>. Consequently, buchapine has the structure of 3-( $\alpha$ , $\alpha$ -dimethylallyl)-3( $\gamma$ , $\gamma$ -dimethylallyl)-1,2,3,4-tetrahydroquinoline-2,4-dione.



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Buchapine is the only alkaloid of the 1,2,3,4-tetrahydroquinoline-2,4-dione type with phenyl substituents of different structures in position 3.

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